Dihydrobenzofurans in which $J_{cis-2,3} > J_{trans-2,3}$: Confirmation Using X-Ray **Diffraction Analysis**

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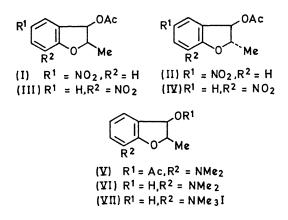
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Summary The validity of the Karplus equation for the assignment of stereochemistry in some substituted 3acetoxy-2-methyl-2,3-dihydrobenzofurans was confirmed by X-ray diffraction analysis.

APPLICATION of the Karplus equation for the assignment of cis-trans stereochemistry has recently been reported to be invalid in some substituted 2,3-dihydrobenzofurans.



In these examples a series of 3-hydroxy (or acetoxy)-2isopropyl-2,3-dihydrobenzofurans were shown to have $trans_{2,3}$ coupling of 6 Hz and $cis_{2,3}$ coupling of 4 Hz.¹

During the preparation of some similar dihydrobenzofurans as potential cholinergic agents, it was necessary to assign cis-trans stereochemistry to the isomers. In view of the reported exception to the Karplus equation, additional studies on structure validation were essential.

The cis-trans isomers† of 5-nitro- and 7-nitro-3-acetoxy-2-methyl-2,3-dihydrobenzofurans (I-IV) were prepared by nitration of the cis-trans mixture of 3-acetoxy-2-methyl-2,3-dihydrobenzofuran; separation of the isomers was accomplished by chromatography and crystallization. The 7-nitro-isomer (III) was reduced to the dimethylaminoderivative, (V) the acetyl group saponified to give (VI), and the resulting amino-alcohol (VI) converted into cis-7-di $methylamino \hbox{-} 3-hydroxy \hbox{-} 2-methyl \hbox{-} 2, 3-dihydrobenzofuran$ methiodide (VII). X-Ray diffraction analyses of (VII) showed the 2-methyl and 3-hydroxy-groups to be cis. Though the hydrogens were not located in this analysis, from the angles of the other substituents at C-2 and C-3 a 19° torsion angle for the hydrogens on C-2 and C-3 of (VII) is a reasonable approximation.

The cis-isomers (I), (III), (V), (VI), and (VII) showed $J_{2,3}$ 6 Hz and the *trans*-isomers (II) and (IV) gave a coupling constant, $J_{2,3}$ 2 Hz. This agrees with the assignments made on the basis of the Karplus equation.²

This work was supported by a predoctoral fellowship (to L.J.P.) and grants from the National Institutes of Health.

(Received, March 18th, 1970; Com. 385.)

† The n.m.r., i.r., and elemental analysis of all compounds in this paper support the assigned structures.

¹ L. H. Zalkow and M. Ghosal, J. Org. Chem., 1969, **34**, 1646; Chem. Comm., 1967, 922. ³ L. M. Jackman and S. Sternhell, "Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry," Pergamon ^a L. M. Jackman and S. Sternhell, "Ap Press, New York, 1969, 2nd edn., p. 281.